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In a tetrahedral molecular geometry, a central atom is located at the center with four substituents that are located at the corners of a tetrahedron. The bond angles are $\cos^{-1}(\frac{1}{\sqrt{3}}) = 109.4712206\dots^\circ \hat{\approx} 109.5^\circ$ when all four substituents are the same, as in methane (CH₄) as well as its heavier analogues. Methane and other perfectly symmetrical tetrahedral molecules belong to point group ...

Tetrahedral molecular geometry - Wikipedia

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Chapter 9

There are many criteria by which superconductors are classified. The most common are: Response to a magnetic field. A superconductor can be Type I, meaning it has a single critical field, above which all superconductivity is lost and below which the magnetic field is completely expelled from the superconductor; or Type II, meaning it has two critical fields, between which it allows partial ...

Superconductivity - Wikipedia

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Anonymous, "An Account of the Battle of Princeton". Pp. 310-312, Pennsylvania Magazine of History and Biography, Vol. 8. Philadelphia: Historical Society of Pennsylvania, 1884. Anonymous, "Council of War At a Board of General Officers Convened at New Windsor 12 June 1781". Pp. 102-103, The Magazine of American history with notes and queries , Vol.III, A. S. Barnes., 1879.

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